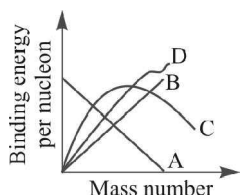
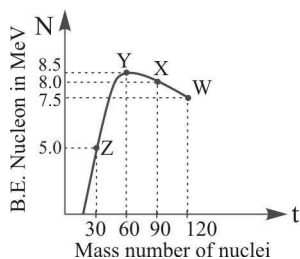


### Diagram Based Questions :

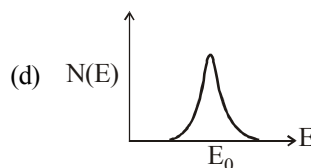
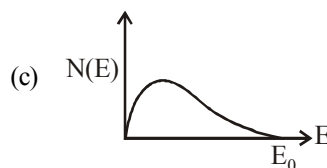
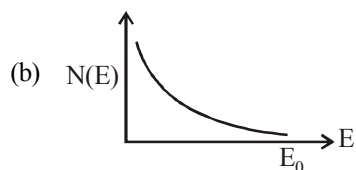
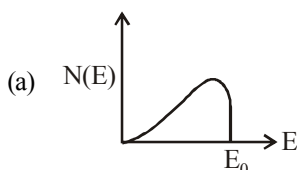
1. Binding energy per nucleon plot against the mass number for stable nuclei is shown in the figure. Which curve is correct?



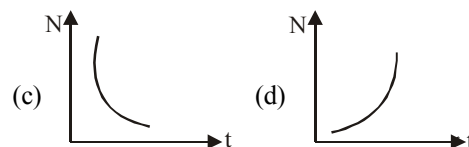
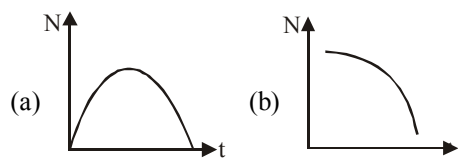
- (a) A (b) B  
(c) C (d) D
2. Binding energy per nucleon versus mass number curve for nuclei is shown in the figure. W, X, Y and Z are four nuclei indicated on the curve. The process that would release energy is



- (a)  $Y \rightarrow 2Z$  (b)  $W \rightarrow X + Z$   
(c)  $W \rightarrow 2Y$  (d)  $X \rightarrow Y + Z$
3. The energy spectrum of  $\beta$ -particles [number  $N(E)$  as a function of  $\beta$ -energy  $E$ ] emitted from a radioactive source is



4. Radioactive element decays to form a stable nuclide, then the rate of decay of reactant is



# Solution

1. (c)
2. (c) Energy is released in a process when total binding energy (BE) of products is more than the reactants. By calculations we can see that this happens in option (c).  
Given  $W = 2Y$   
BE of reactants =  $120 \times 7.5 = 900$  MeV  
BE of products =  $2 \times (60 \times 8.5) = 1020$  MeV.
3. (c) The range of energy of  $\beta$ -particles is from zero to some maximum value.
4. (c) No. of nuclide at time  $t$  is given by

$$N = N_0 e^{-\lambda t}$$

Where  $N_0$  = initial nuclide

thus this equation is equivalent to  $y = ae^{-kx}$

thus correct graph is

